

ALD Capabilities at Penn State

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Penn State Nanofab

4 ALD tools: 2 thermal & 2 plasma



Savannah 200 thermal



KJ Lesker
thermal 150LE



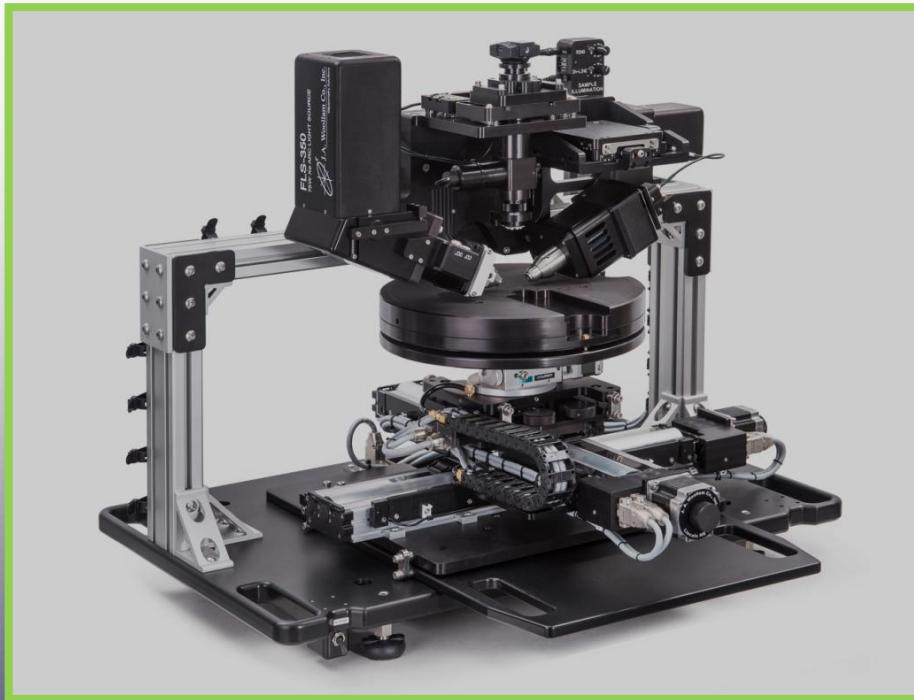
KJ Lesker
Cluster 150LX



KJ Lesker
Plasma
150LX



Thickness measurement and mapping



Woollam 2000X focused beam Ellipsometer



ALD processes on Savannah 200

	T (°C)	300	250	200	170	150	120	110	100	80	60	50
Regular chamber (6.5 mm tall)	Al ₂ O ₃	0.85 /1.65	1.0 /1.65	1.03 /1.65	1.04 /1.63	1.08 /1.65			0.93 /1.55	0.96		0.82 /1.60
	HfO ₂		1.14 /2.07	0.95 /2.02				1.16				
	TiO ₂					0.44 /2.29				0.7		
	ZnO ₂			1.35 /2.11			1.83			1.49	1.28	
	ZrO ₂			0.97 /2.05								
	Ta ₂ O ₅			0.55 /1.72								
	AZO											
Tall chamber (13 mm tall)	Al ₂ O ₃			1.03 /1.64								



ALD processes on Lesker LE



	T (°C)	300	250	200	150	120	110	100
process	Al ₂ O ₃	0.84 /1.61	0.88 /1.63	0.93 /1.66	0.96 /157		0.75 /1.57	0.76 /1.62
	HfO ₂		0.97 /2.11	0.98 /2.05	1.10 /2.04			
	TiO ₂				0.43 /2.00			
	Ta ₂ O ₅	0.47 /1.75		0.53 /1.99	0.63 /1.98	0.78 /1.93		
Long water pulse(1.6s)	Al ₂ O ₃			1.06	1.03 /1.61		0.78 /1.6	0.93 /1.66
EXPO	Al ₂ O ₃			1.22				
	Ta ₂ O ₅			0.65 /2.03				



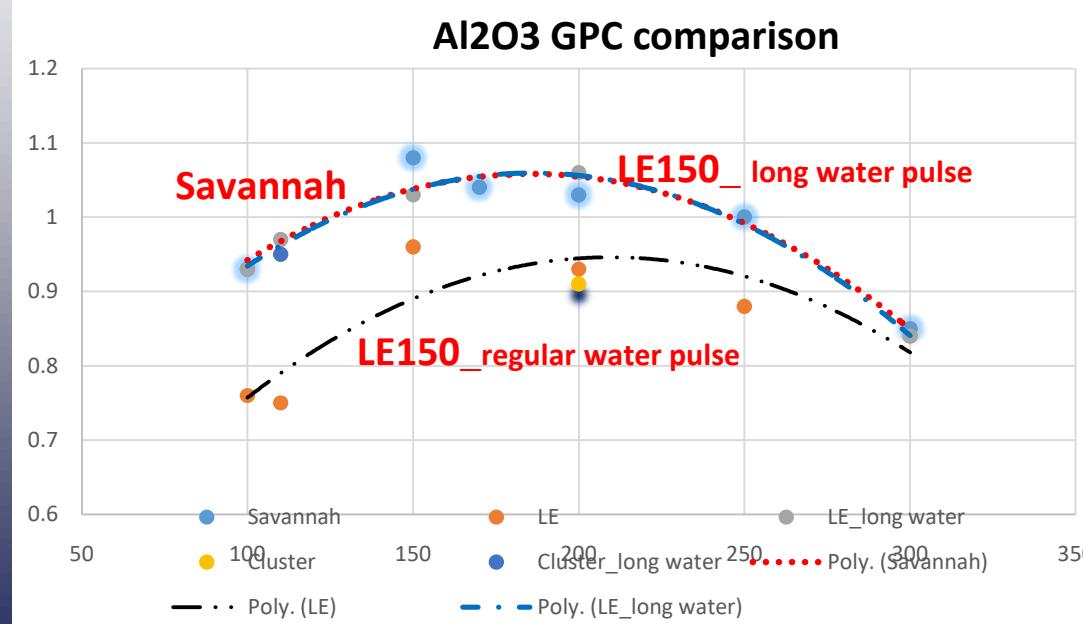
ALD processes on Lesker Cluster LX150

	T (°C)	350	300	250	200	120	110	100
Thermal	Al ₂ O ₃		0.85 /1.64		0.91 /1.65			
Long water pulse(1.6s)	Al ₂ O ₃						0.95 /1.6	
Plasma process	Al ₂ O ₃		0.84 /1.63		1.06 /1.66			
	HfO ₂				1.09 /2.02	1.23 /2.00		
	SiO ₂		0.96 /1.41					
	TiN	0.40 /165 μΩ cm	0.41 /379 μΩ cm					
	AlN			0.58 /1.96	0.69 /1.93			
	GaN							1.14 /1.86
	Pt		0.6-0.8 /11 cyl (Nucl delay)					



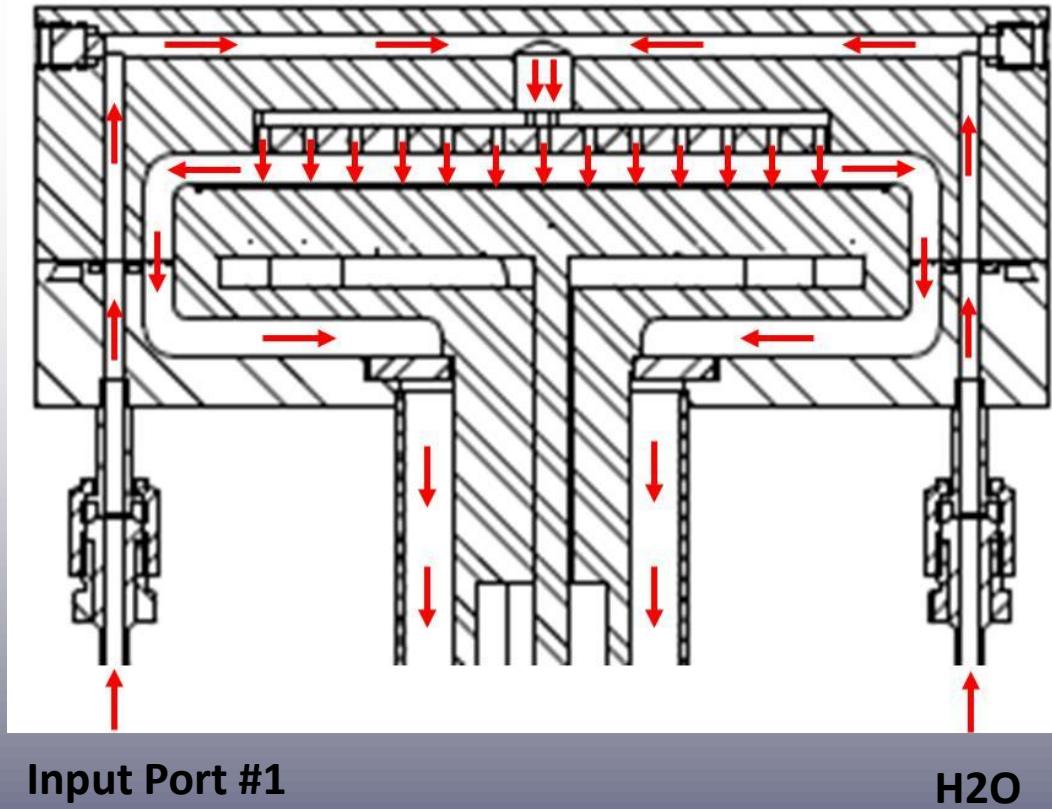
Comparison between different systems

	T (°C)	Savannah thermal	LE thermal	Cluster thermal	Cluster Plasma
Al ₂ O ₃	300	0.85	0.84	0.85	0.84
	200	1.03	0.93		1.06
HfO ₂	200	0.95	0.98		1.09
Ta ₂ O ₅	200	0.55	0.53		
TiO ₂	150	0.44	0.43		





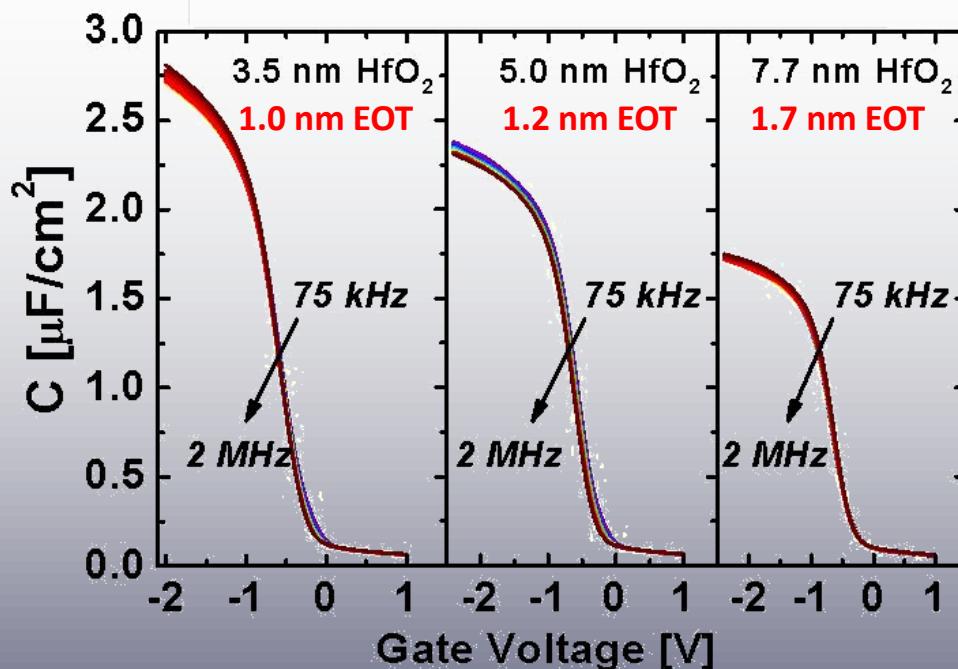
Different gas flow design from Savannah



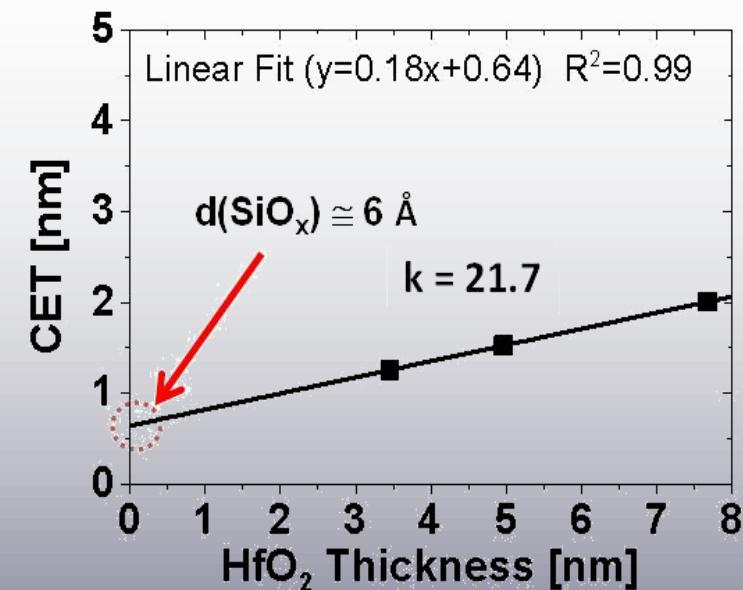
- Inactive gas flow is continuous, viscous and laminar – precursor gas/vapor periodically pulsed through
- Net direction of flow is perpendicular to substrate surface

CV measurement of MOSCAP for HfO₂

with thicknesses of 1.0, 1.2 & 1.7 nm EOT



Steep slope & small frequency dispersion
indicative of very low interface trap density
($D_{it} < 10^{11} \text{ cm}^{-2} \text{ eV}^{-1}$)

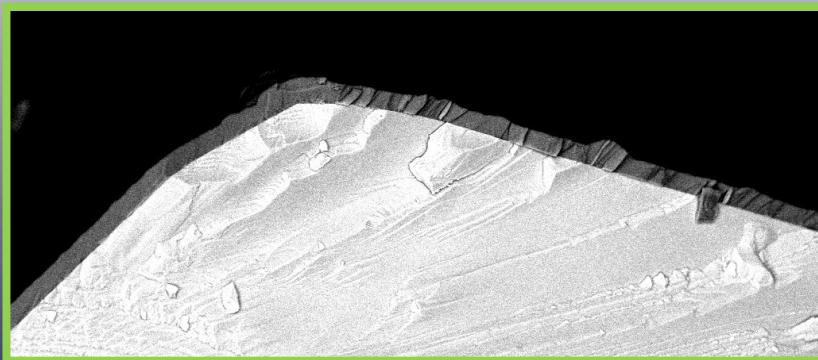


Indicating a high dielectric constant
($k \sim 22$) and thin interfacial oxide
thickness ($\sim 0.6 \text{ nm}$) required for
aggressive device scaling (1 nm
EOT)



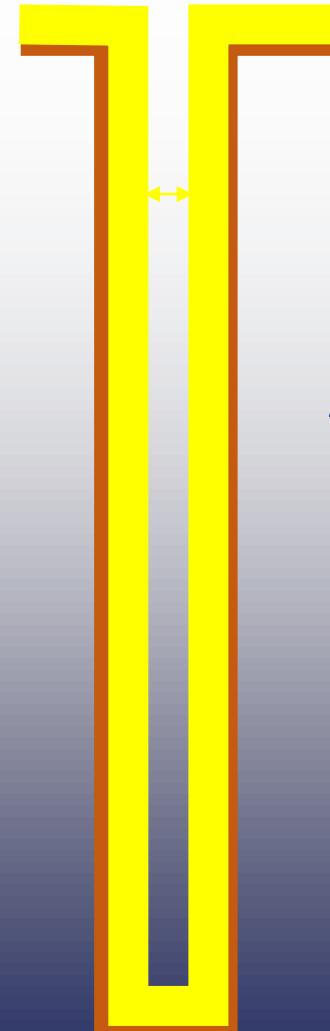
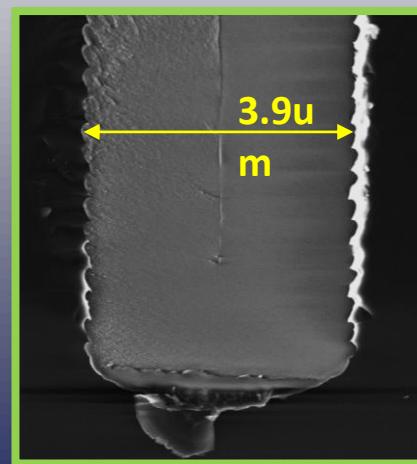
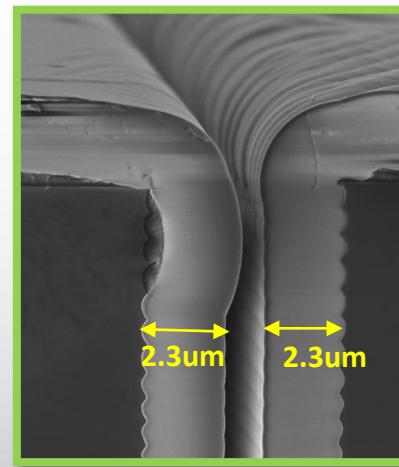
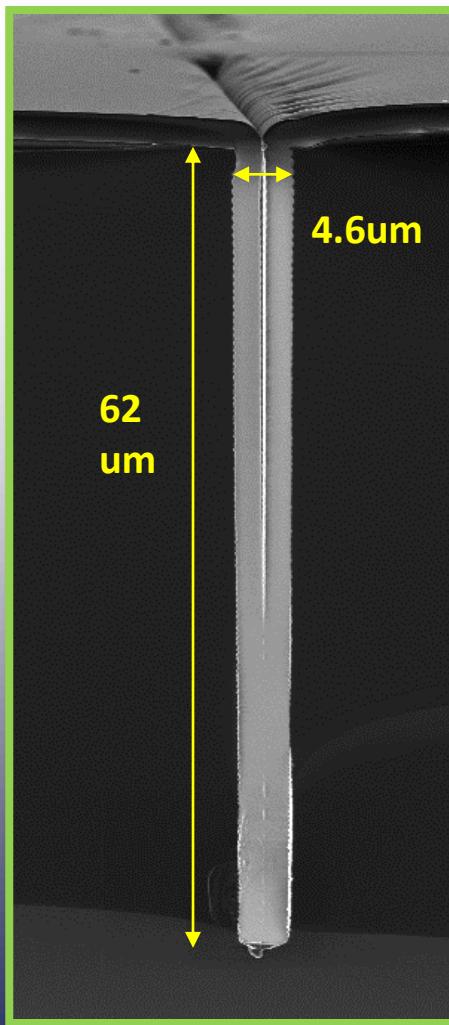
How thick a cylinder of 25 gram TMA can deposit Al₂O₃ film on Savannah?

8 um



1 um Al₂O₃ on fiber

Regular thermal Al₂O₃ process on Savannah



Aspect ratio can
be larger than
100:1!

Coating multiple wafers on Savannah

Al₂O₃ 170C

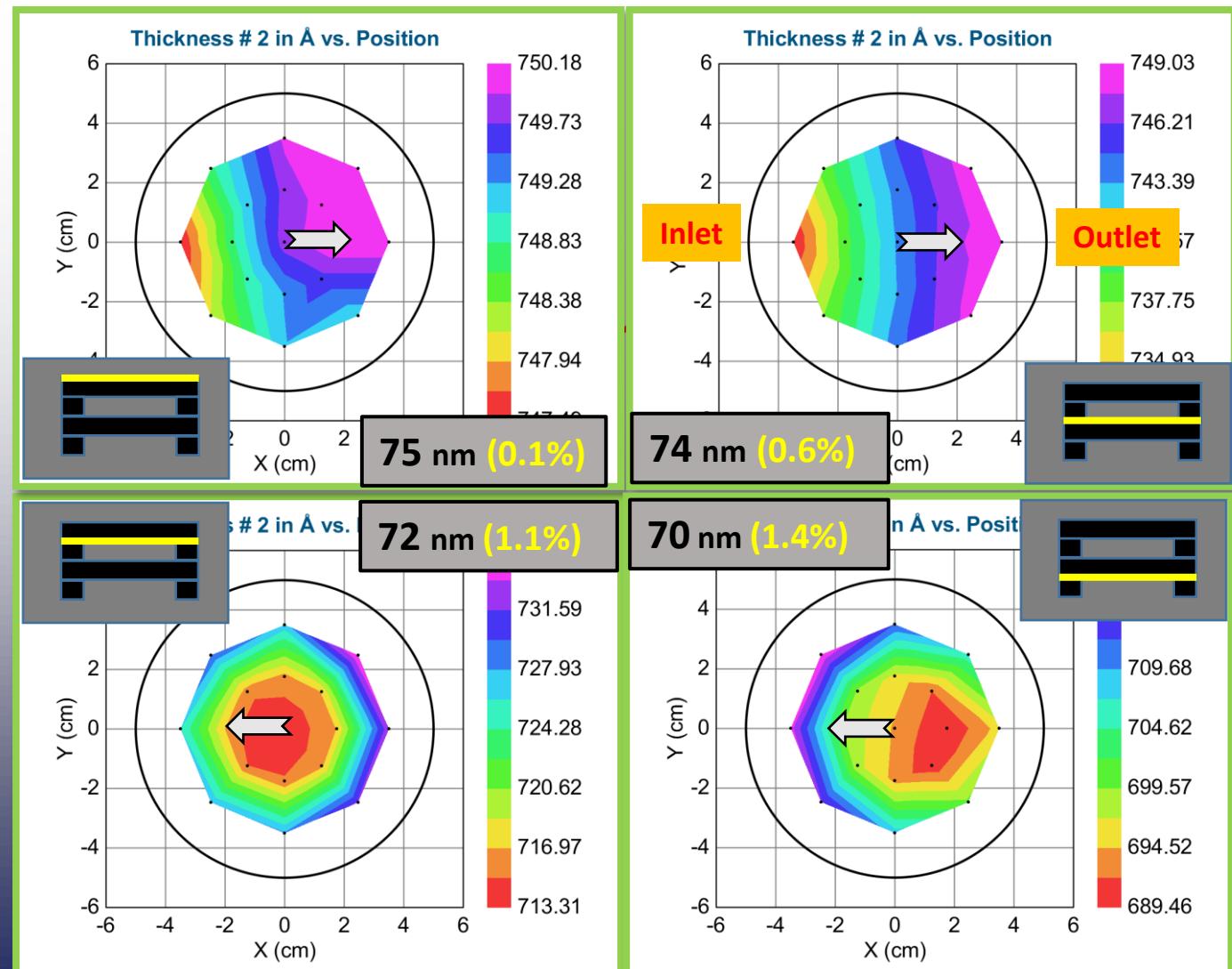
700 cycles

2X 4"

double side polished

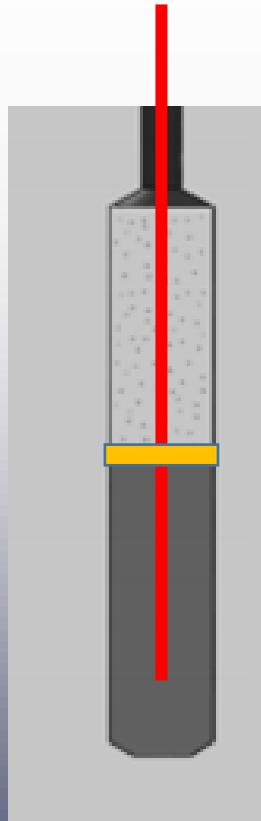
Si wafers

Si spacer: 0.7mm





Precursor overtemperature damage



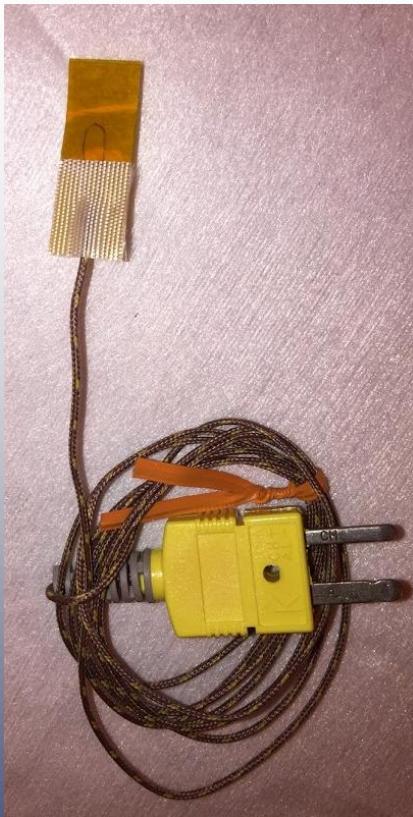
Tetrakis(dimethylamido)zirconium, TDMAZr,

Tetrakis(dimethylamino)tin(IV)

Pentakis(dimethylamido)tantalum

**TDMAZr forms a yellowish thin crust on top when overheated
The precursor can be reused when poking through the crust**

Better temperature control using self adhesive thermocouple & heating tape



Stable Ta₂O₅ process

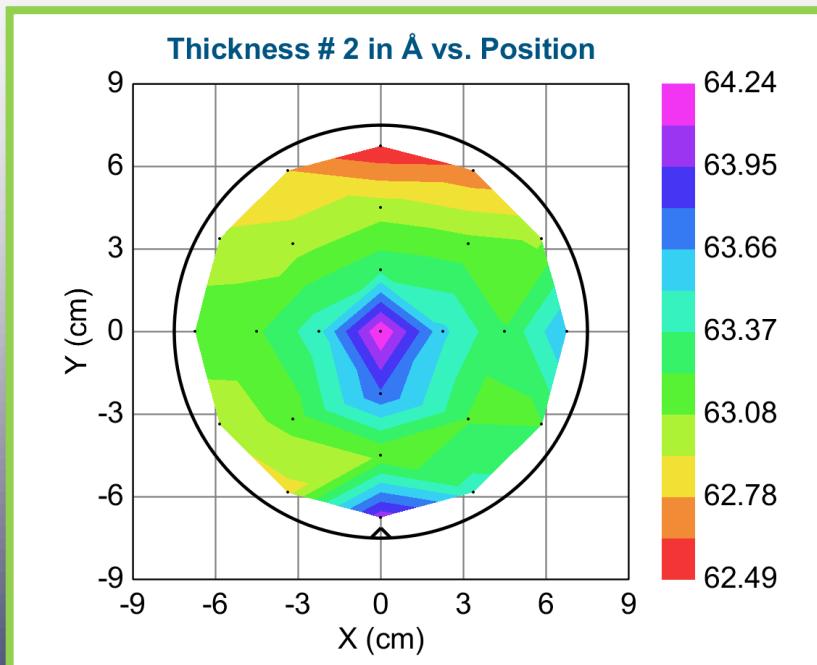
Ta₂O₅ static 200C 100 cycles

GPC: 0.63 Å/cycle

Substrate: 150 mm Si

Native oxide: 1.15 nm

180911



$\langle d \rangle: 63.23 \pm 0.4 \text{ Å}$ (0.6% uniformity)

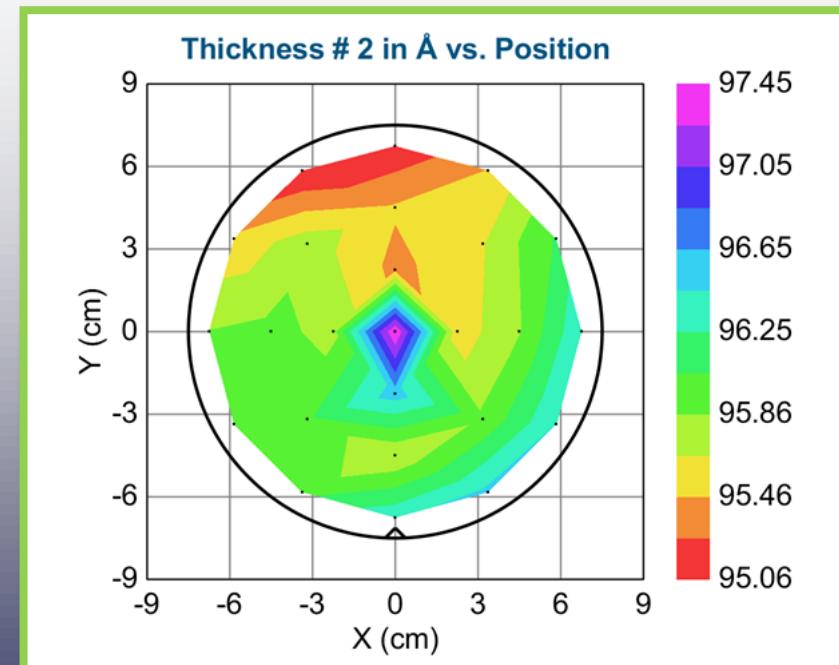
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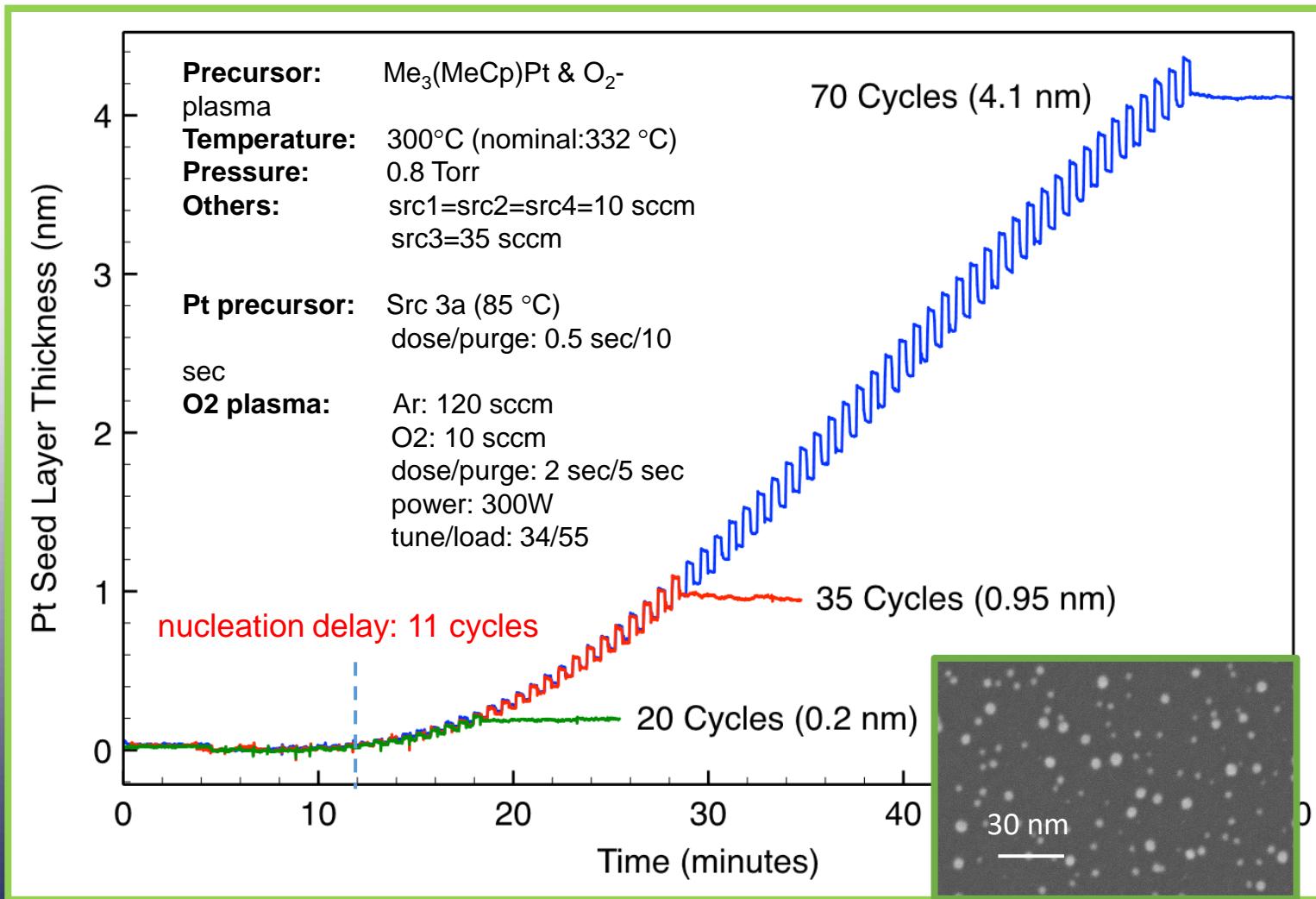
190605



$\langle d \rangle: 95.9 \pm 0.5 \text{ Å}$ (0.5% uniformity)



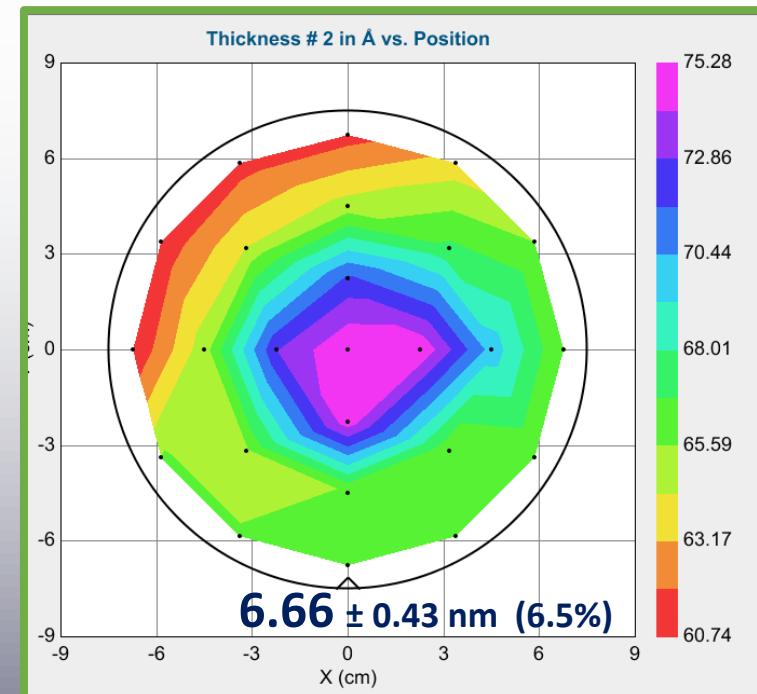
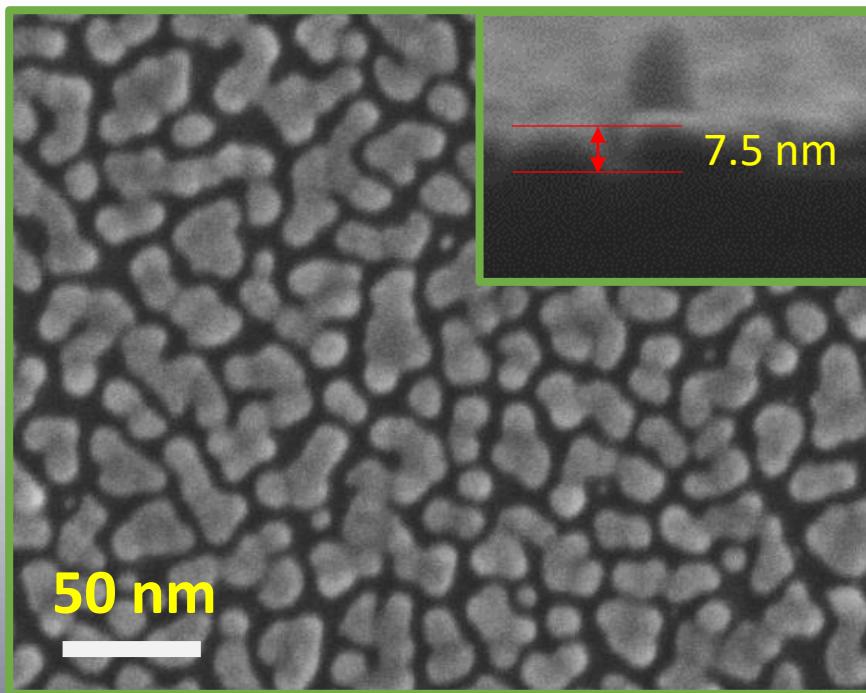
PEALD Pt process





Thickness measurement of Pt film

100 cycles of Pt deposition



Layer Commands: [Add](#) [Delete](#) [Save](#)

Include Surface Roughness = [OFF](#)

Layer # 2 = [Gen-Osc](#) Thickness # 2 = **117.96 Å** (fit)

[Add Oscillator](#) [Show Dialog](#) Fast Gaussian Calc = [ON](#)

Einf = **2.387** (fit)

UV Pole Amp. = **0.0000** UV Pole En. = **11.000**

IR Pole Amp. = **0.0000**

[Fit All](#) [Clear All](#) [Add Amp.](#) [Add Br.](#) [Add En.](#)

1: Type = [Drude\(RT\)](#) Resistivity (Ohm-cm)1 = **0.00023159** (fit) Scat. Time (fs)1 = **0.518** (fit)

2: Type = [Lorentz](#) Amp2 = **0.640632** (fit) Br2 = **0.1750** (fit) En2 = **3.126** (fit)

3: Type = [Lorentz](#) Amp3 = **4.595097** (fit) Br3 = **3.2924** (fit) En3 = **7.030** (fit)

4: Type = [Lorentz](#) Amp4 = **3.111312** (fit) Br4 = **2.6162** (fit) En4 = **5.071** (fit)

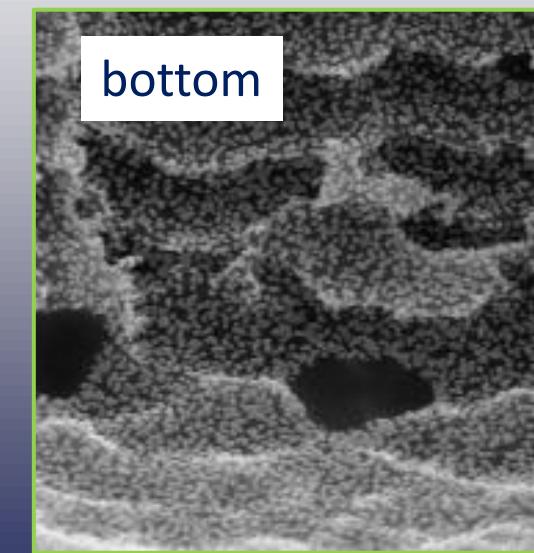
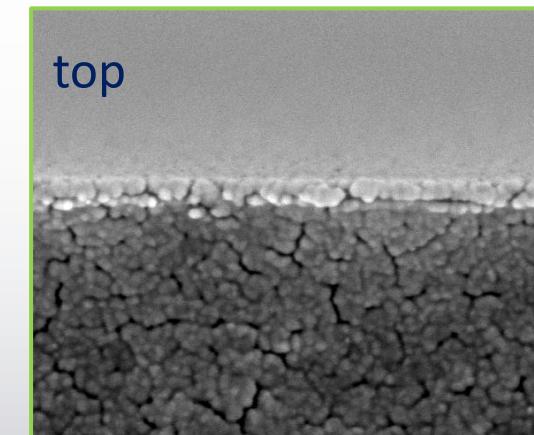
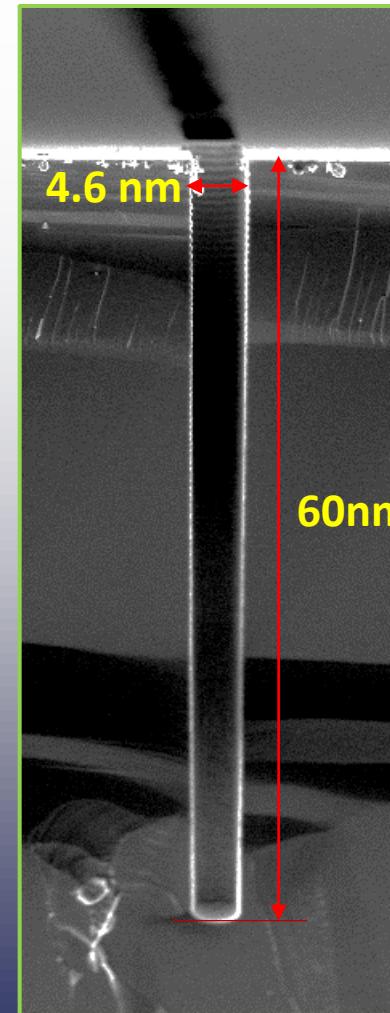
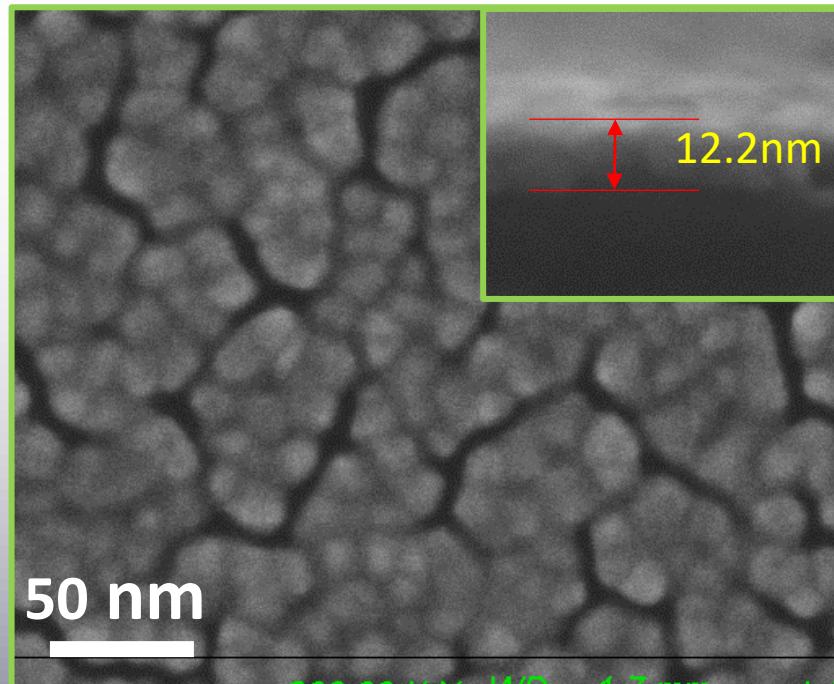
Layer # 1 = [NTVE_JAW](#) Native Oxide = **11.80 Å**

Substrate = [Si_JAW](#)

Drude model

good for conductive film: Pt & TiN

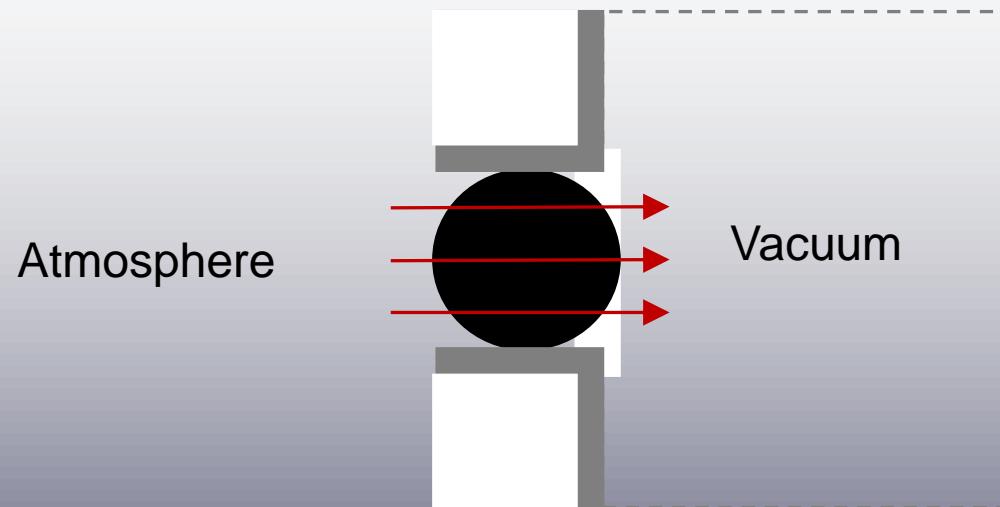
200 cycle Pt deposition on Si & trenches



Bottom has less deposition than top



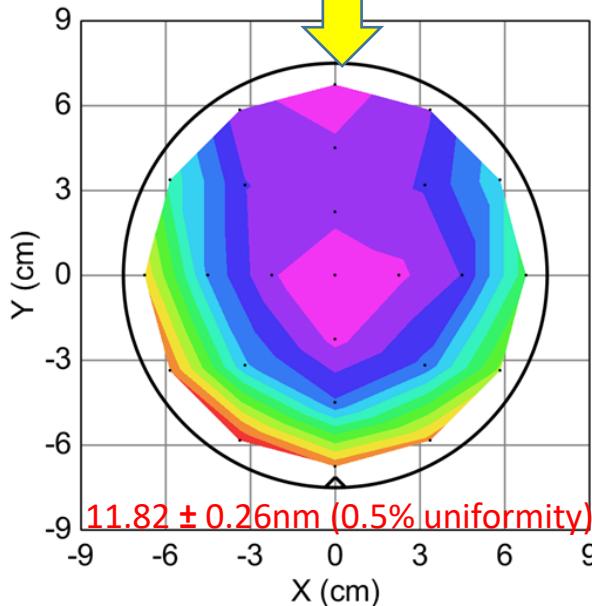
Does Permeation through O-ring affect ALD process?



TiN thickness mapping shows gradient related to air permeation

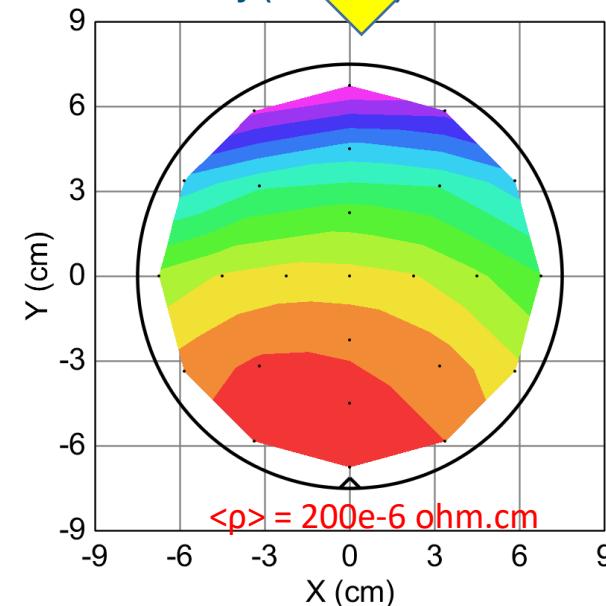
Air permeation

Thickness # 2 Å vs. Position



Air permeation

Resistivity (Ohm·cm) vs. Position



975W, 300 cycles, 300C

GPC: **0.394 Å/cycle**

Substrate: 150 mm Si

Native oxide: 1.18 nm

TiCl₄ + H₂:N₂ plasma (3:1 ratio)

Process Pressure: ~1.150 Torr

Substrate heater temperature: 400C (~321C using carrier)

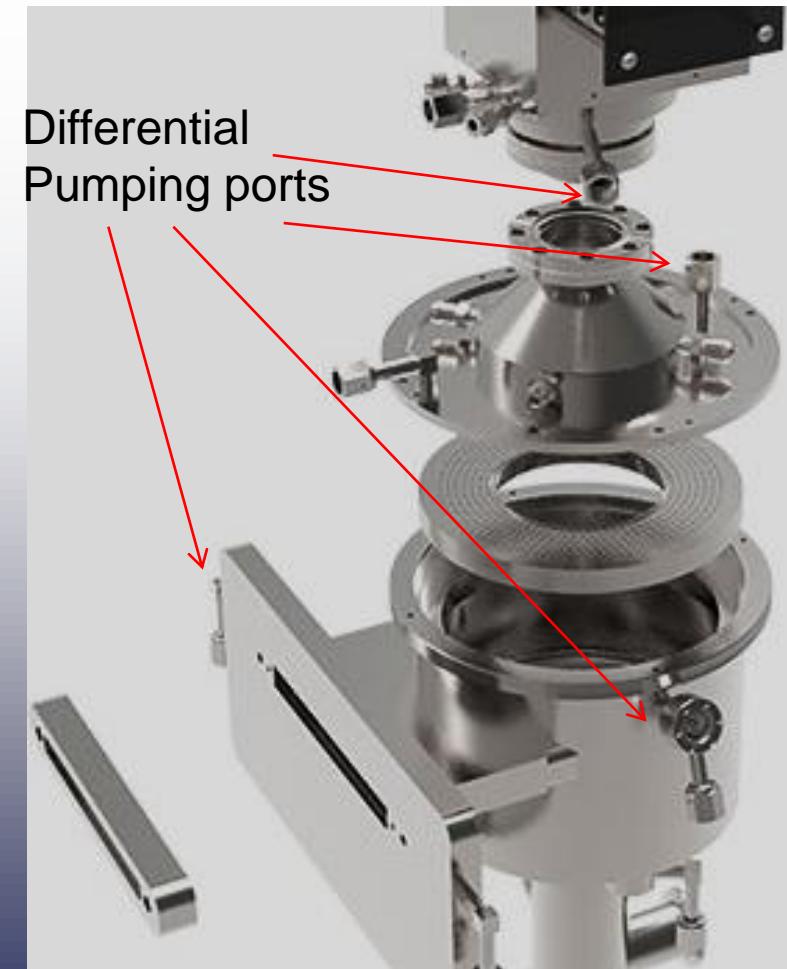
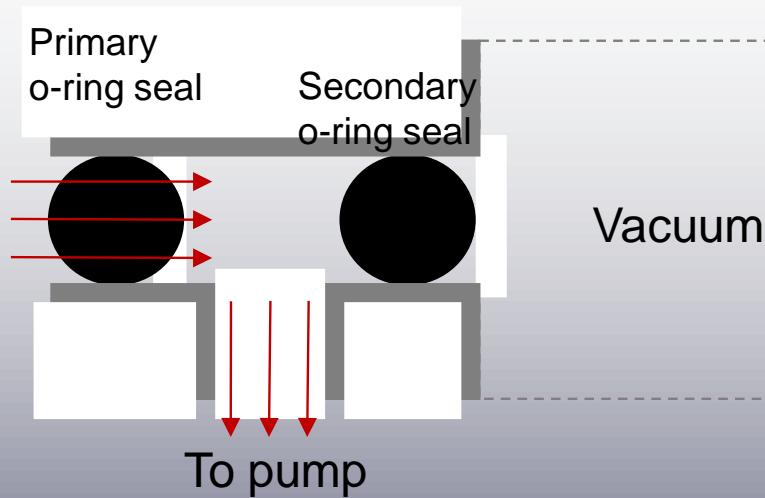
TiCl₄ Dose/purge: 0.1 sec/3 sec

Plasma Dose/purge: 9 sec/5 sec

Cycle Time: ~17 seconds

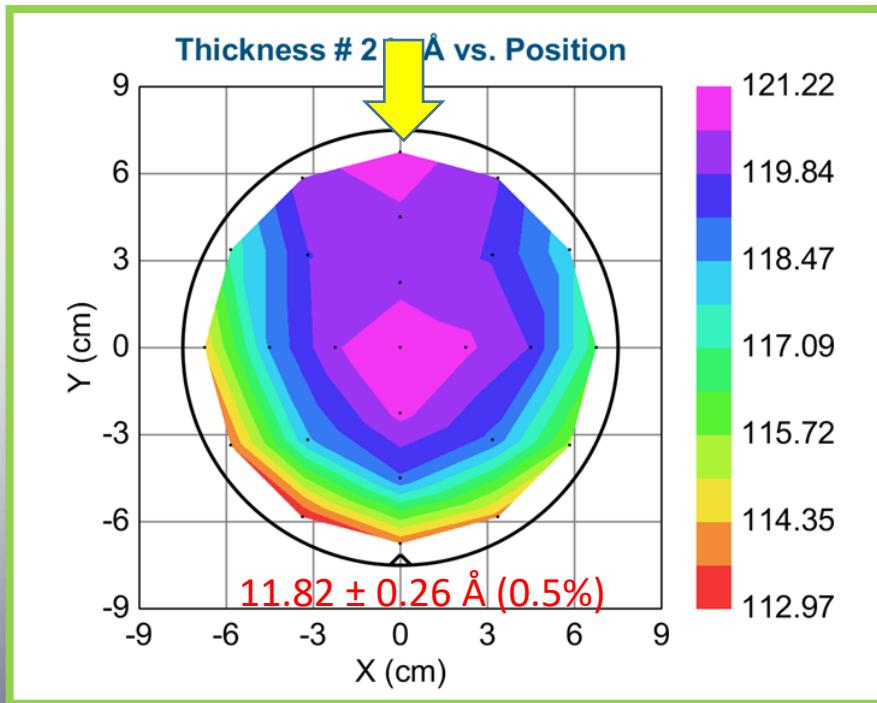


Solution: Double O-rings & differential pumping



Effect of differential pumping on TiN growth

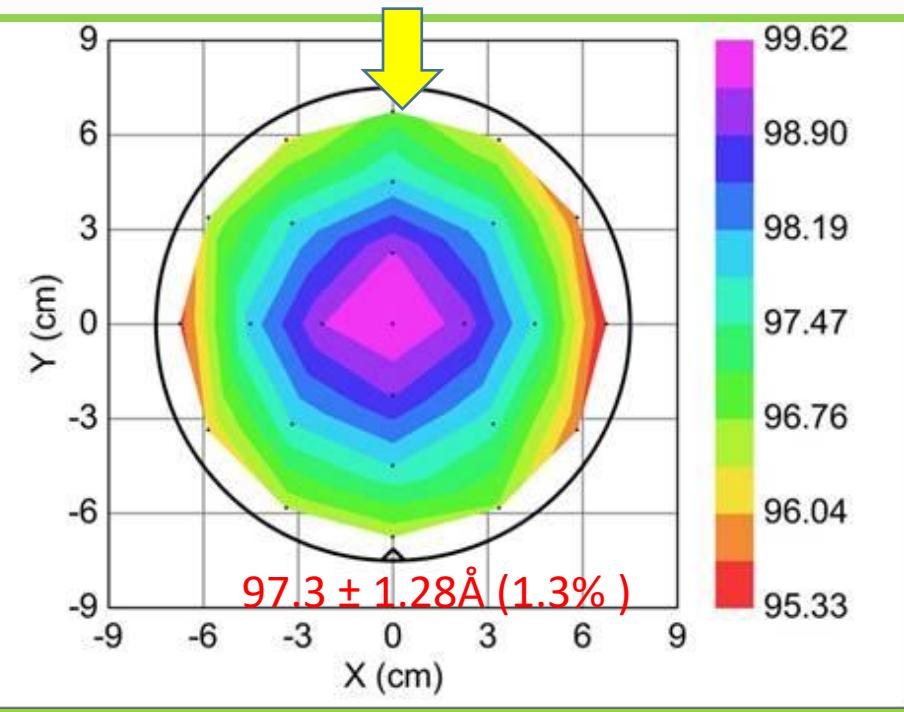
Air permeation



GPC: 0.39 Å/cycle

w/o differential pumping
(Cluster LX150)

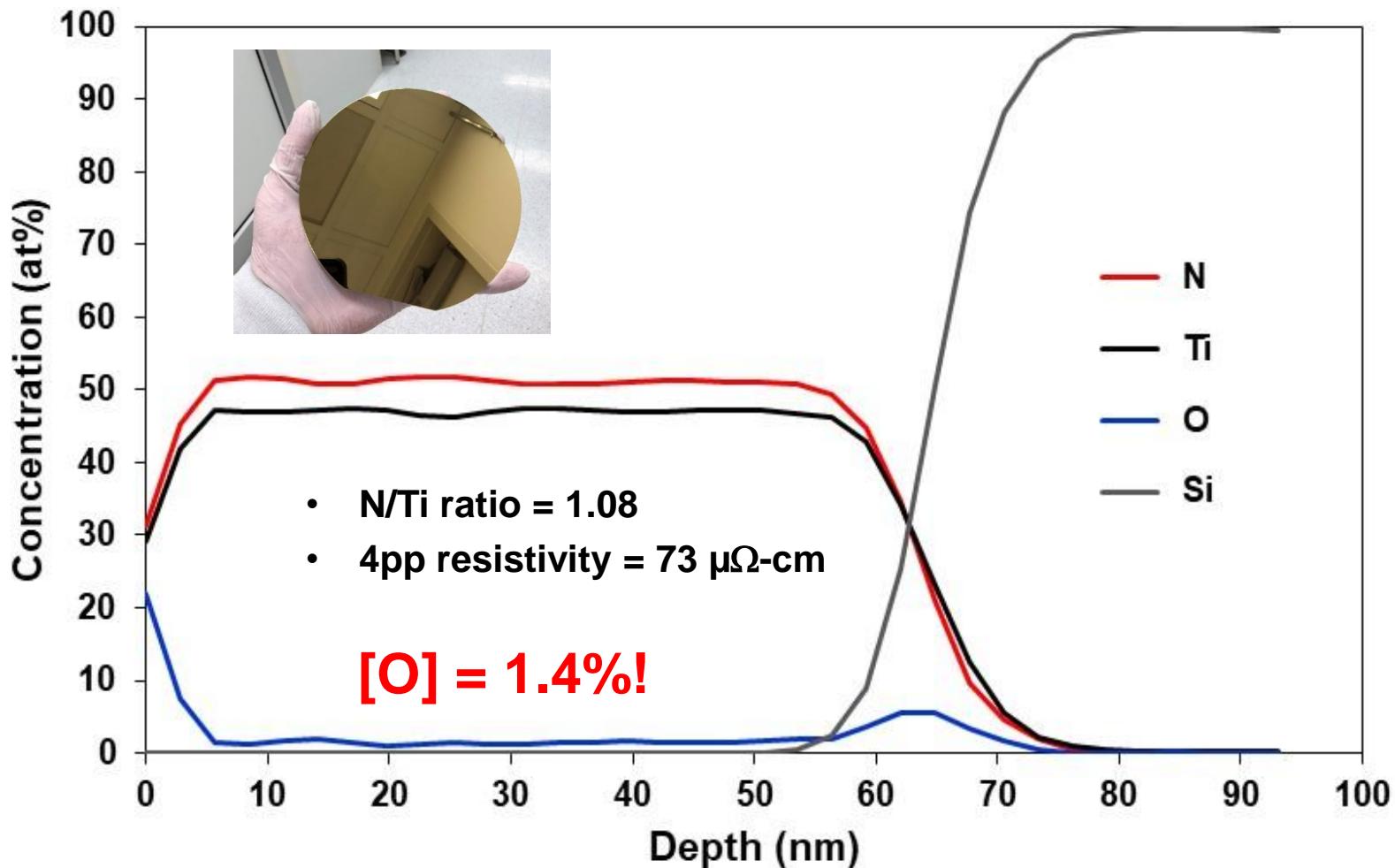
Air permeation



GPC: 0.32 Å/cycle

w/ differential pumping
(LX150)

XPS confirms extremely low O in TiN

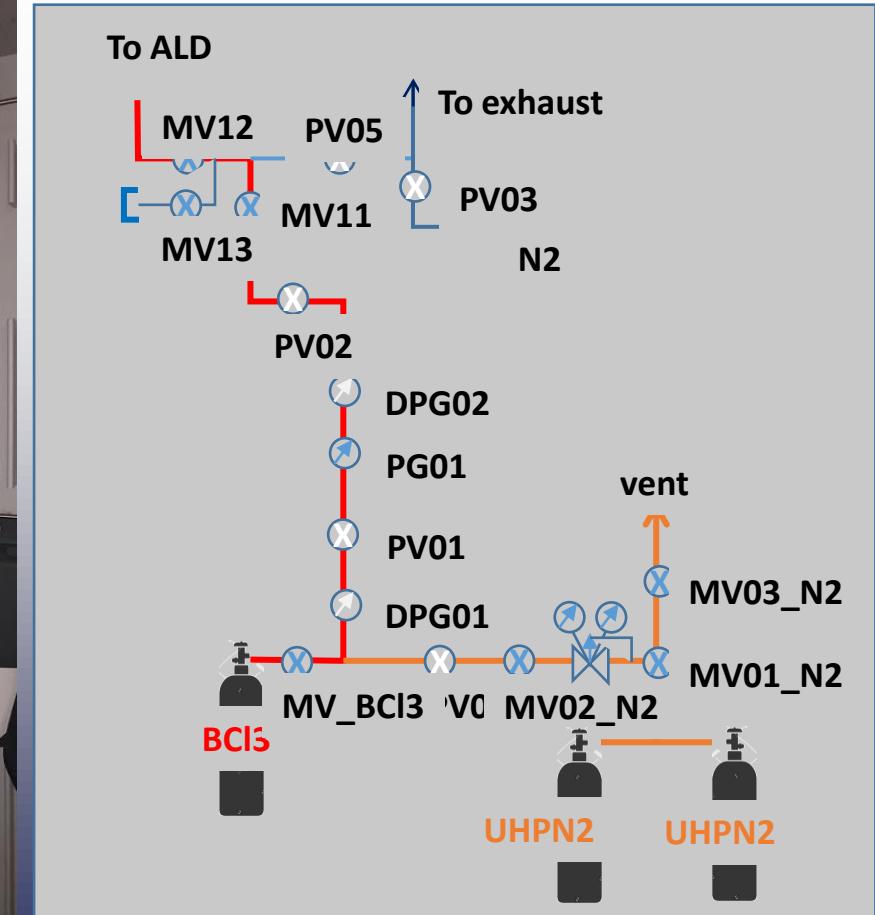


BCl₃ gas delivery system for BN deposition

BCl₃: toxic & corrosive

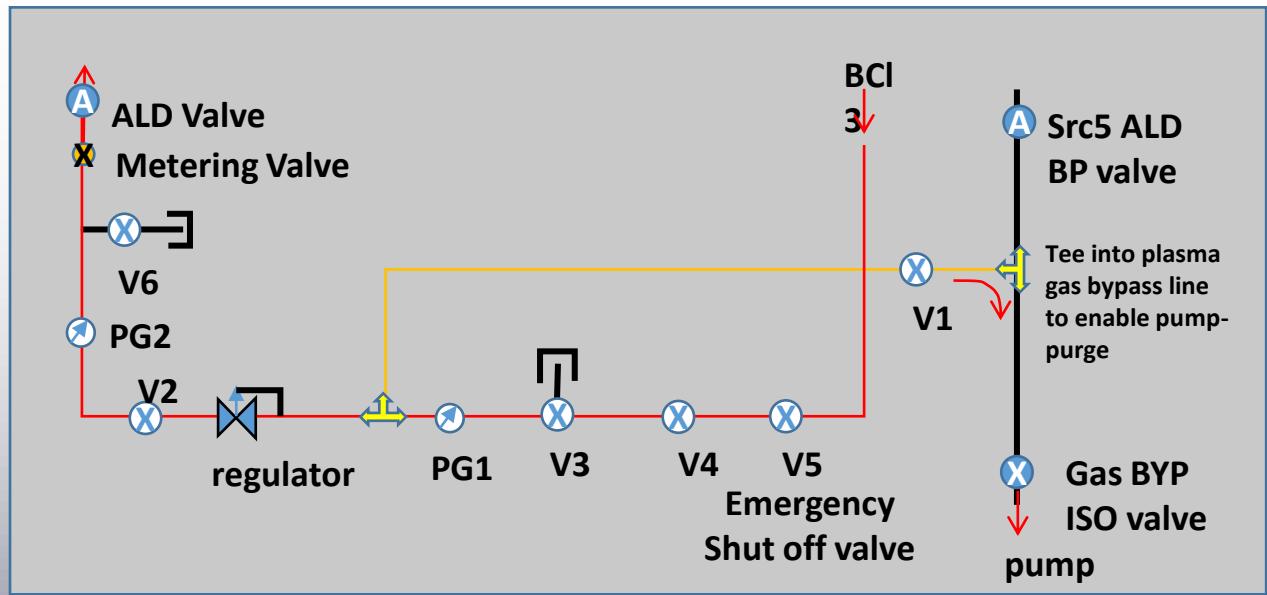
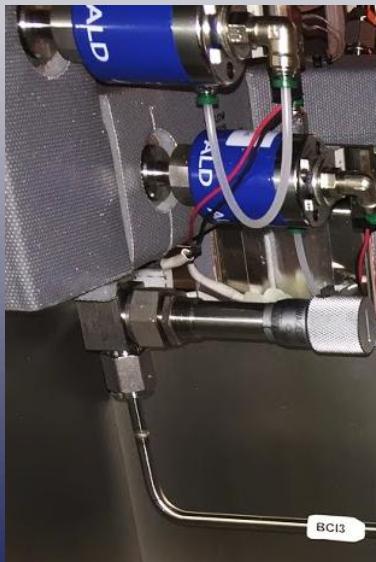


Gas delivery in basement

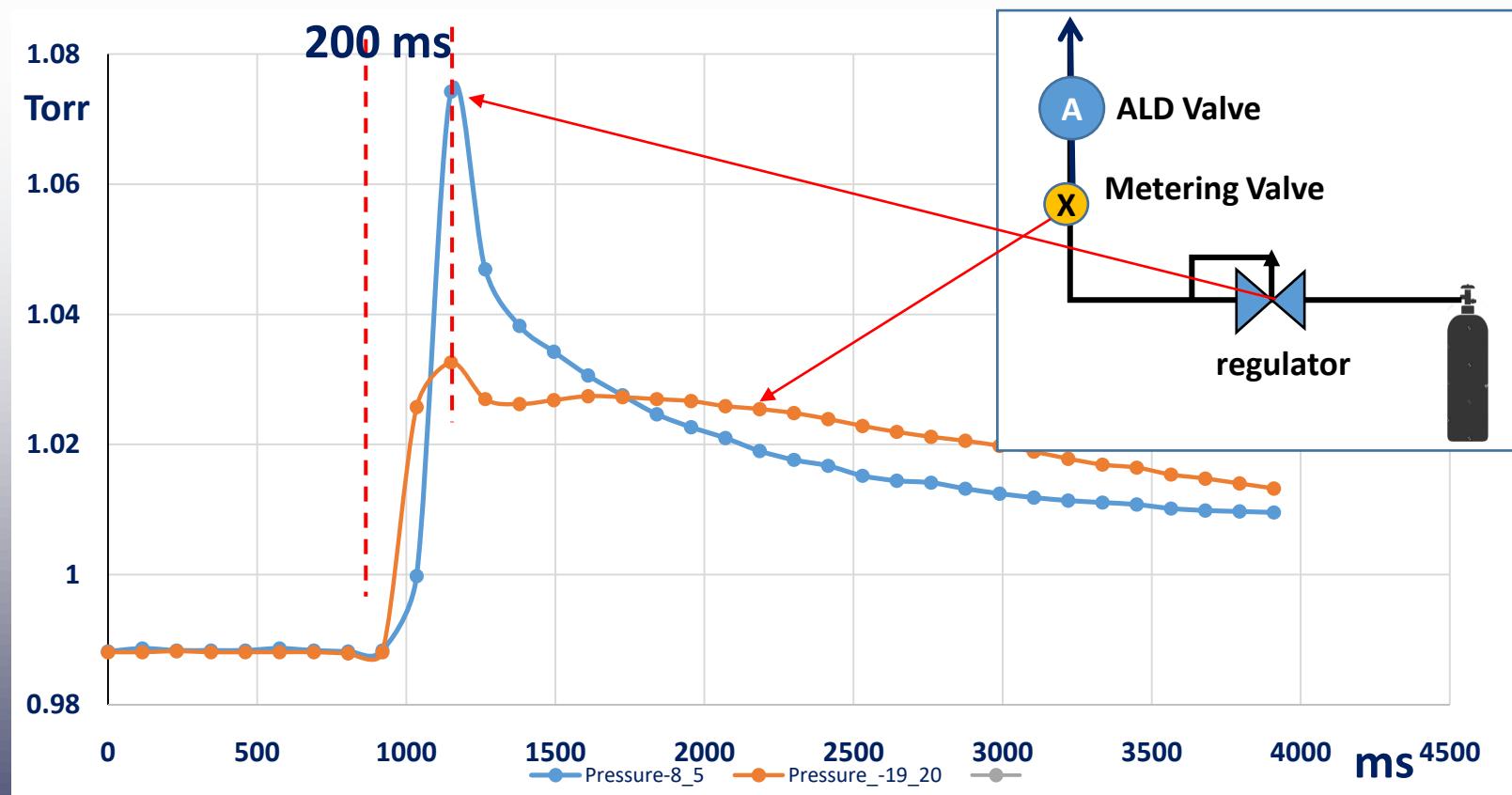


BCl₃ gas delivery system for BN deposition

Gas delivery on tool



Pulse height tuning using N₂



Peak height is controlled by regulator

Steady state flow is controlled by metering valve

Thank you!